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IL21 – The Evolution from FLPs to Electrophilic Phosphonium Cations and Hydrodefluorination and C–F Functionalization

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Suitably selected combinations of main group Lewis acids and bases constitute frustrated Lewis pairs (FLPs) were shown to activate hydrogen. This has allowed the development of unprecedented metal-free catalytic hydrogenations of C=N bonds, anilines, *N*-heterocycles, olefins, polyaromatic hydrocarbons and most recently ketones and aldehydes.¹ This conceptual advance has been furthered leading to the development of electrophilic phosphonium cations (EPCs).² These species prove to be highly Lewis acidic species that can effect CF bond hydrodefluorination catalysis. In addition, these Lewis acids are useful for hydrosilylations,³ dehydrocouplings,⁴ ketone deoxygenation⁵ and in FLP hydrogenations.⁶ More recently, we have extended the reactivity of EPCs to effect the catalytic Fiedel-Crafts arylation of CF₃ groups⁷ as well as benzyl fluorides. Such reactivity has little precedent and provides a new and highly variable approach to organic derivatizations. The implications of these findings for applications of main group species in catalysis is considered in this lecture.

References

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